## DISCIPLINE DESCRIPTION

1. Information on the study programme

1.1 Academic institution	UNIVERSITY OF ORADEA
1.2 Faculty	FACULTY OF ENVIRONMENTAL PROTECTION
1.3 Department	Food Engineering
1.4 Field of study	Food Engineering
1.5 Cycle of study	BACHELOR
1.6 Study programme/Qualification	Control and Expertise of Food Products / Engineer

2. Information on the discipline

2.1 Name of discipline			Physics II					
2.2 Course holder Lecturer Phys. Eng. Alin Cristian TEUŞDEA, PhD								
2.3 Seminar/Laboratory/Project			Lec	cture	r Phys. Eng. Alin Cristian	TEU	JŞDEA, PhD	
holder								
2.4 Year of study	1	2.5 Semeste	er	2	2.6 Type of evaluation	Ex	2.7 Regime of discipline	C

<sup>(</sup>C) Compulsory; (O) Optional; (E) Elective

3. Total estimate time (hours per semester of didactic activities)

3.1 Number of hours per	2	out of	2	out of which 3.3	0/ 2/0	
week		which: 3.2		seminar/laboratory/project		
		course				
3.4 Total hours in the	56	out of	28	out of which 3.6	0/28/0	
curriculum		which: 3.5		seminar/laboratory/project		
		course				
Time allotment						
Study assisted by manual, course support, bibliography and notes						
Additional documentation in the library/ on specialised electronic platforms and in the field						
Preparation of seminars/laboratories/ topics/reports, portfolios and essays						
Tutorship						
Examinations						
Other activities						

3.7 Total hours of individual study	40
3.9 Total hours per semester	96
3.10 Number of credits	4

**4. Prerequisites** (where appropriate)

-	or rerequisites (	-ppi-spiiate)
	4.1 curriculum	-
	4.2 competences	-

**5. Conditions** (where appropriate)

5.1. related to course	<ul> <li>The course is based on oral presentation with video projector, notebook with MS PowerPoint software, Adobe Reader, Internet access. Students can ask questions and have the obligation to follow the course schedule.</li> <li>During the course students will not be present with open mobile phones.</li> </ul>
5.2. related to seminar/laboratory/ project	<ul> <li>For practical work, it is mandatory to prepare (study) each practical work a week before.</li> <li>Each student will conduct an individual activity with the equipment and laboratory materials that will be</li> </ul>

	completed by performing the calculations described in the
	laboratory guide.
•	During laboratory work, students are not allowed to make
	telephone calls within the laboratory.

6. Spe	6. Specific competences acquired						
Professional competences	<ul> <li>Identify and appropriately use the main laws and physical principles in a given context: Apply physics principles and laws in solving theoretical or practical problems under qualified assistance.</li> <li>To develop the ability to explain phenomena in food engineering as a consequence of applying physics laws in the context of the complexity of food systems</li> <li>To develop the ability to use lab techniques necessary for food engineering designing experimental design, obtaining experimental data, analyzing and interpreting them and formulating conclusions</li> <li>To apply the knowledge in the physics field of both in concrete situations in related fields and in experiments, using standard laboratory equipment.</li> </ul>						
Transversal competences	<ul> <li>To demonstrate preoccupation for professional development through the use of practical thinking skills, engineering</li> <li>To participate in scientific projects         Acquiring / completing the information needed to assimilate the content of disciplines in food engineering     </li> </ul>						

7. Objectives of discipline (coming from the specific competences acquired)

1. Objectives of discipline (coming from	Objectives of discipline (coming from the specific competences acquired)				
7.1 General objective	<ul> <li>Acquiring specific the language and the notions related to physical phenomena that arise in the field of food engineering.</li> </ul>				
7.2 Specific objectives	<ul> <li>Acquiring specific language</li> <li>Acquiring insights regarding the physical phenomena that arise in the field of food engineering</li> <li>Interpretation of physics equations and their correct application in experiments</li> <li>Performing experimental measurements, processing and interpreting the results</li> <li>Identifying applications specific to the field of food engineering in which the physical phenomena were studied</li> </ul>				

### 8. Content\*/

8.1 Course	Methods of teaching	No. of hours/ Remarks
Geometric Optics.  Light dispersion. Types of optical prisms with applications in optical systems and spectroscopy.	systematic exposure, conversation, problem- solving,	2
Applications in refractometry (Abbe refractometer) in order to determine the concentrations of some solutions.	systematic exposure, conversation, problem- solving,	2
Spherical diopters, flat diopters.	systematic exposure, conversation, problem- solving,	2

Slim lenses.	systematic exposure, conversation, problem- solving,	2
The thick lens.	systematic exposure, conversation, problem- solving,	2
Centric optical systems.  Design of centric optical systems (C.O.S.) from two thin lenses	systematic exposure, conversation, problem- solving,	2
Design of centric optical systems (C.O.S.) of two spherical diopters, thick lens	systematic exposure, conversation, problem- solving,	2
Optical instruments.  Human eye, magnifying glass, microscope, telescope.  Objectives and eyepieces of optical instruments.  Applications in biophysics.	systematic exposure, conversation, problem- solving,	2
Media of optics with variable refractive index	systematic exposure, conversation, problem- solving,	2
Electromagnetic Optics. General notions of electromagnetism - Maxwell's equations. Electromagnetic waves.	systematic exposure, conversation, problem- solving,	2
The study light polarization. Applications of polarizer with penumbra in biophysics.	systematic exposure, conversation, problem- solving,	2
Interference of light. The condition of interference. Interference of light by dividing the wave front. Interference devices by dividing the wave front.	systematic exposure, conversation, problem- solving,	2
Biophysics applications in food preservation Physical Principles of Atmospheric Conservation: High Pressures; with magnetic field; microwave and high frequency streams.	systematic exposure, conversation, problem- solving,	2
Physical principles of conservation by thermal methods: with ionizing radiation; with ultraviolet radiation; by ohmic heating.	systematic exposure, conversation, problem- solving,,	2
<b>Obs:</b> * Oral exposures, videoprojector presentations, simulations		

#### **Bibliography**

- 1. Creangă Ileana, Fizică (I), Ed. Matrix Rom, București, 2005.
- 2. Creangă Ileana, Fizică (II), Ed. Matrix Rom, București, 2014.
- 3. Boer A., Optică, Ed. Matrix Rom, București, 2006.
- 4. Demșoreanu B., Mecanică teoretică, Tipografia Universității Timișoara, 1991.
- 5. Irina Nicoară, Introducere în optică, Tipografia Universității Timișoara, 1990.
- 6. I. Luminosu, Fizică, Tipografia Universității Tehnice "Politehnica" Timișoara, 1991.
- 7. Alin C. Teuşdea, Fizică generală prin aplicații practice, Ed. Universității din Oradea, ediția a 2-a, 2012, ISBN 978-606-10-0778-3; 53(075.8).
- 8. Alin C. Teuşdea, Elemente de biofizică în tehnică, Curs, 2012.
- 9. Lungu, C., 2002, Principii generale de conservare a produselor alimentare, Universitatea "Dunărea de Jos" IDD, Galați.

8.2 Seminar	Methods of teaching	No. of hours/ Remarks
8.3 Laboratory		

Perpendicular oscillations composition - Lissajous 2D and 3D figures.  Specific heat of the dough determination by the calorimetric method  Specific heat of bananas determination by the calorimetric method  Specific heat of bananas determination by the calorimetric method  Relative air humidity determination by means of hygrometer.  Wheatstone Bridge. Measure specific resistance.  Determining the focal length of a convergent lens by the direct method.  Study of light intensity distribution with photoelectric cell I.  Elements of refractometry. Determination of refractive index of liquids and Brix index with Abbe refractometer  The concentration of a solution determination with the calibration curve of the refractive index of polarimetry. The concentration of aqueous solutions of optically active substances determination with polarimetre with penumbra.  The concentration of an aqueous solution of optically active substances determination with polarimetre with penumbra.  The concentration of an aqueous solution of optically active substance determination with polarimetre with penumbra.  The concentration of an aqueous solution of optically active substance determination with polarimetre with penumbra.  The concentration of an aqueous solution of optically active substance determination with polarimetre with penumbra. solving.  The concentration of an aqueous solution of optically active substance determination with polarimetre with penumbra. solving.  The concentration of an aqueous solution of optically active substance determination with polarimetre with penumbra. solving.  The concentration of an aqueous solution of optically active substance determination with polarimetre with penumbra. solving.  The concentration of an aqueous solution of optically active substance determination with polarimetre with penumbra. solving.  The concentration of an aqueous solution of optically active substance determination with polarimetre with penumbra. solving.  The concentration of an aqueous solution of optically active sub	NTSM and the rules of operation of the laboratory	systematic exposure,	2
calorimetric method  Specific heat of bananas determination by the calorimetric method  Relative air humidity determination by means of hygrometer.  Wheatstone Bridge. Measure specific resistance.  Wheatstone Bridge. Measure specific resistance.  Determining the focal length of a convergent lens by the direct method.  Study of light intensity distribution with photoelectric cell I.  Elements of refractometry. Determination of refractive index of liquids and Brix index with Abbe refractometer  The concentration of a solution determination with the calibration curve.  Elements of polarimetry. The concentration of aqueous solutions of optically active substance determination with polarimeter with penumbra.  The concentration of an aqueous solution of optically active substance determination with healibration curve.  Design of Compact Centric Optical Systems:  2. Two thin lens system (microscope).  Conversation, problem-solving, systematic exposure, conversation, problem-solving, systematic expo	Perpendicular oscillations composition - Lissajous	conversation, problem-	2
calorimetric method  Relative air humidity determination by means of hygrometer.  Wheatstone Bridge. Measure specific resistance.  Wheatstone Bridge. Measure specific resistance.  Determining the focal length of a convergent lens by the direct method.  Study of light intensity distribution with photoelectric cell I.  Elements of refractometry. Determination of refractive index of liquids and Brix index with Abbe refractometer  The concentration of a solution determination with the calibration curve of the refractive index.  Elements of polarimetry. The concentration of aqueous solutions of optically active substances determination with polarimeter with penumbra.  The concentration of an aqueous solution of optically active substances determination with the calibration curve.  Design of Compact Centric Optical Systems:  1. The thick lens;  2 conversation, problem-solving,  systematic exposure, conversation, problem-solving,  systematic exposure, conversation, problem-solving,  ysystematic exposure, conversation, problem-solving,  conversation, problem-solving,  2 /experiment  2 /experiment  2 /experiment		conversation, problem- solving,	_
hygrometer.   conversation, problem-solving, systematic exposure, conversation, problem-solving, systematic exposure, conversation, problem-solving, solving, solving,   2 / experiment	calorimetric method	conversation, problem- solving,	/experiment
Determining the focal length of a convergent lens by the direct method.  Study of light intensity distribution with photoelectric cell I.  Elements of refractometry. Determination of refractive index of liquids and Brix index with Abbe refractometer  The concentration of a solution determination with the calibration curve of the refractive substances determination with polarimeter with penumbra.  Elements of polarimetry. The concentration of aqueous solutions of optically active substances determination with polarimeter with penumbra.  Design of Compact Centric Optical Systems:  2. Two thin lens system (microscope).  Design of Compact Centric Optical Systems:  2. Two thin lens system (microscope).		conversation, problem- solving,	/experiment
the direct method.  Study of light intensity distribution with photoelectric cell I.  Elements of refractometry. Determination of refractive index of liquids and Brix index with Abbe refractometer  The concentration of a solution determination with the calibration curve of the refractive substances determination of an aqueous solution of optically active substance determination with the calibration curve.  Design of Compact Centric Optical Systems:  1. The thick lens;  Conversation, problem-solving, systematic exposure, conversation	<u> </u>	conversation, problem- solving,	/experiment
Elements of refractometry. Determination of refractive index of liquids and Brix index with Abbe refractometer  The concentration of a solution determination with the calibration curve of the refractive index.  Elements of polarimetry. The concentration of aqueous solutions of optically active substances determination with polarimeter with penumbra.  The concentration of an aqueous solution of optically active substances determination with polarimeter with penumbra.  The concentration of an aqueous solution of optically active substance determination with polarimeter with penumbra.  The concentration of an aqueous solution of optically active substance determination with the calibration curve.  Design of Compact Centric Optical Systems:  1. The thick lens;  Design of Compact Centric Optical Systems:  2. Two thin lens system (microscope).		conversation, problem-	
refractive index of liquids and Brix index with Abbe refractometer  The concentration of a solution determination with the calibration curve of the refractive index.  Elements of polarimetry. The concentration of aqueous solutions of optically active substances determination with polarimeter with penumbra.  The concentration of an aqueous solution of optically active substance determination with polarimeter with penumbra.  The concentration of an aqueous solution of optically active substance determination with the calibration curve.  Design of Compact Centric Optical Systems:  1. The thick lens;  Design of Compact Centric Optical Systems:  2. Two thin lens system (microscope).  conversation, problem-solving,  systematic exposure, conversation, problem-solving,  2 /experiment  /experiment  /experiment		conversation, problem-	_
the calibration curve of the refractive index.  Elements of polarimetry. The concentration of aqueous solutions of optically active substances determination with polarimeter with penumbra.  The concentration of an aqueous solution of optically active substance determination with the calibration curve.  Design of Compact Centric Optical Systems:  1. The thick lens;  Design of Compact Centric Optical Systems:  2. Two thin lens system (microscope).    Conversation, problem-solving,   Conversat	refractive index of liquids and Brix index with Abbe	conversation, problem-	_
aqueous solutions of optically active substances determination with polarimeter with penumbra.  The concentration of an aqueous solution of optically active substance determination with the calibration curve.  Design of Compact Centric Optical Systems:  1. The thick lens;  Design of Compact Centric Optical Systems:  2. Two thin lens system (microscope).    Conversation, problem-solving,   Conversation, problem-solving,		conversation, problem-	_
optically active substance determination with the calibration curve.  Design of Compact Centric Optical Systems:  1. The thick lens;  Design of Compact Centric Optical Systems:  2 conversation, problem-solving,  Systematic exposure, conversation, problem-solving,  Systematic exposure, conversation, problem-solving,  2 / simulation  2 / simulation	aqueous solutions of optically active substances	conversation, problem-	<del>-</del>
1. The thick lens;  Design of Compact Centric Optical Systems:  2. Two thin lens system (microscope).  conversation, problem-systematic exposure, conversation, problem-solving,  solving,  / simulation // simulation	optically active substance determination with the	conversation, problem-	
2. Two thin lens system (microscope). conversation, problem-solving,		conversation, problem-	<u> </u>
8.4 Project		conversation, problem-	<del>-</del>
	8.4 Project	-	-

#### **Bibliography**

- 1. Creangă Ileana, Fizică (I), Ed. Matrix Rom, București, 2005.
- 2. Creangă Ileana, Fizică (II), Ed. Matrix Rom, București, 2014.
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- 6. I. Luminosu, Fizică, Tipografia Universității Tehnice "Politehnica" Timișoara, 1991.
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# 9. Corroboration of discipline content with the expectations of the epistemic community, professional associations and representative employers from the field corresponding to the study programme

#### 10. Evaluation

Type of activity	10.1 Evaluation criteria	10.2 Evaluation methods	10.3 Share in the final grade	
10.4 Course	Written exam	Written paper	50%	
10.5 Seminar	-	-	-	
10.6 Laboratory	Practical colloquium	Executing a practical	50%	
		work/experimet		
10.7 Project	-	-	-	
10.8 Minimum standard of performance:				
Each of the two components of the final mark must be passed with a minimum grade of 5 (five).				

Date of completion	Signature of <b>course</b> holder**	Signature of seminar laboratory/project holder **	
	Alin Cristian Teuşdea, PhD, ateusdea@uoradea.ro	Alin Cristian Teuşdea, PhD, ateusdea@uoradea.ro	
Date of approval in the Department		Signature of the Head of Department	
		Lecturer Eng. Adrian V. Timar, PhD	
		Dean signature	
		Prof. Eng. Ioan Chereji, PhD	

<sup>\*</sup> The content, respectively the number of hours allocated to each course / seminar / laboratory / project will be detailed during the 14 weeks of each semester of the academic year.